

# Competition Between Cavitation and Flow in Amorphous Solids

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postdoc  
opportunity!

JOHNS HOPKINS

WHITING SCHOOL  
*of* ENGINEERING

# Bulk Metallic Glasses

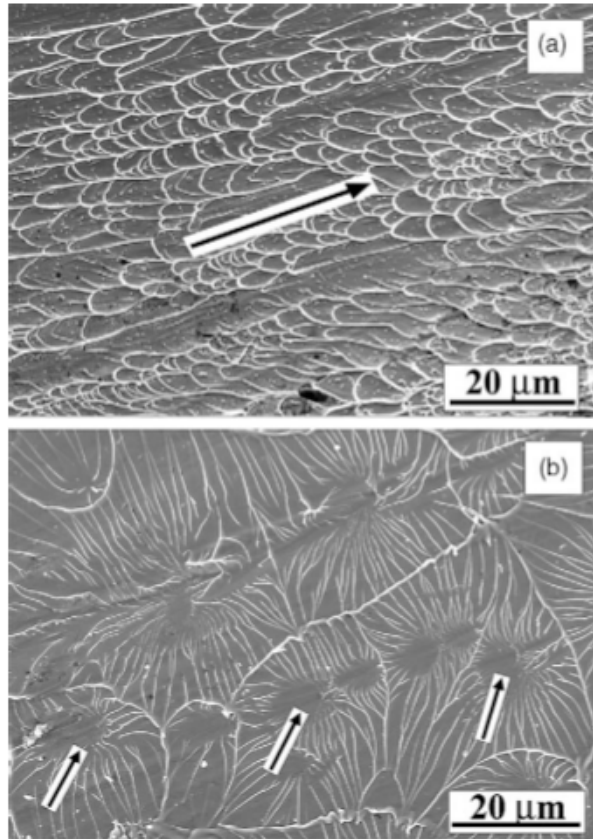


Douglas Hoffman, E-Science News 1998

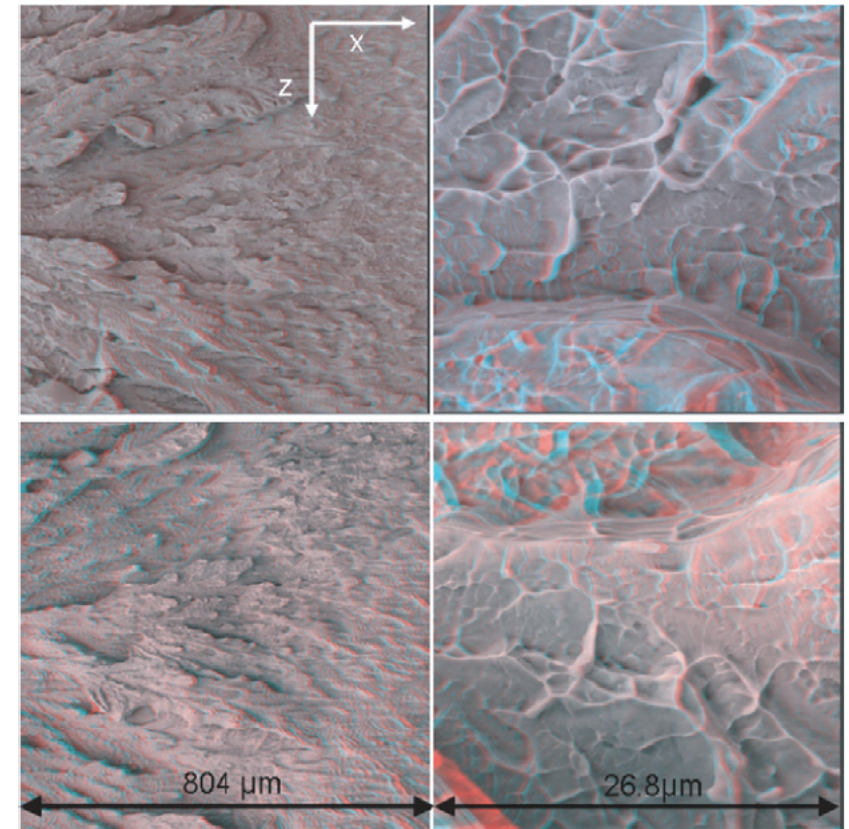
- Unique material with promising properties, but disordered structure.
- What can molecular dynamics simulation tell us about the mechanical response of these materials?

# Fracture in Metallic Glass

- Phenomenology of fracture in metallic glass



Compressive and Tensile Fracture in  
 $\text{Zr}_{59}\text{Cu}_{20}\text{Al}_{10}\text{Ni}_8\text{Ti}_3$   
Zhang, Eckert and Schultz  
Acta Materialia, 51, pp. 1167 (2003)

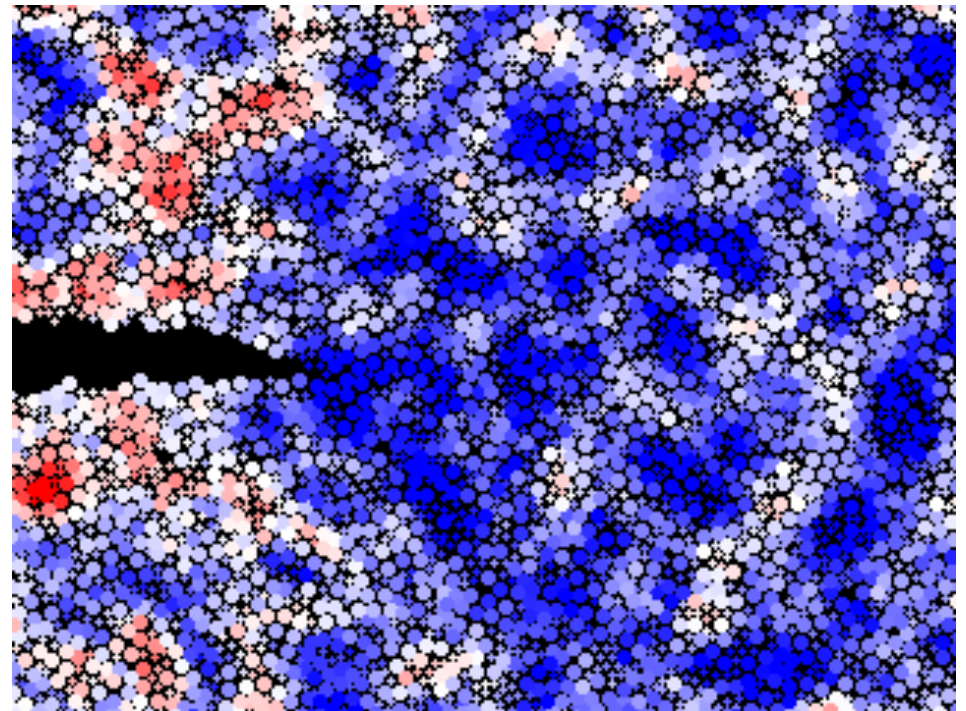
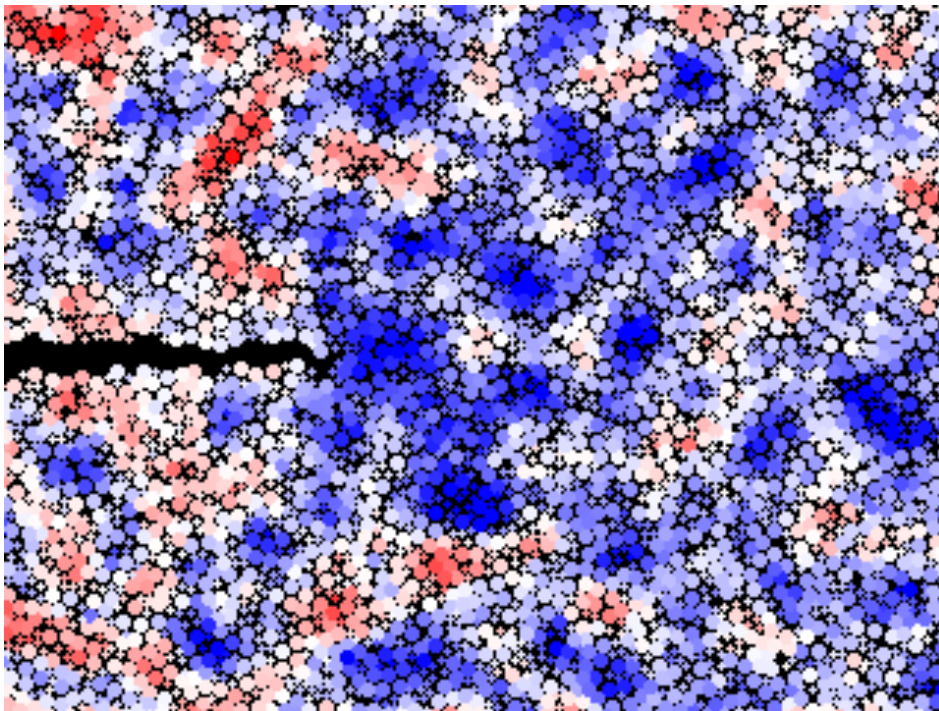


Tensile fracture in Vitreloy I  
Bouchaud, Boivin, Pouchou, Bonamy, Poon and  
Ravichandran  
Europhysics Letters, V83, pp. 66006 (2008)



# MD Simulations of Fracture

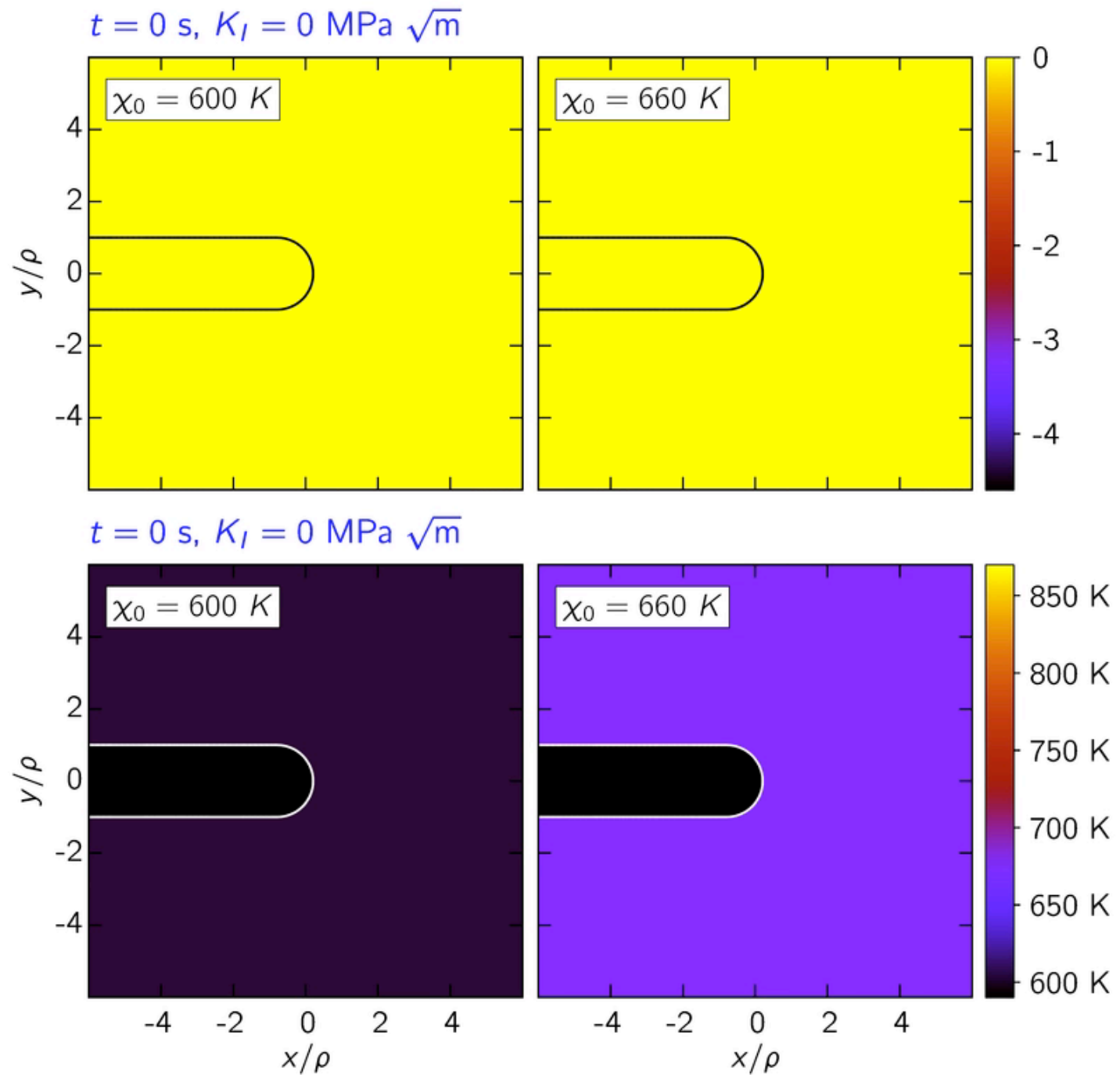
- Simple model of a polydisperse system interacting with pairwise potential
- Both "ductile" and "brittle" simulations exhibit cavitation ahead of crack tip.



ML Falk, PRB, V60, 7062 (1999)

# STZ Crack Tip

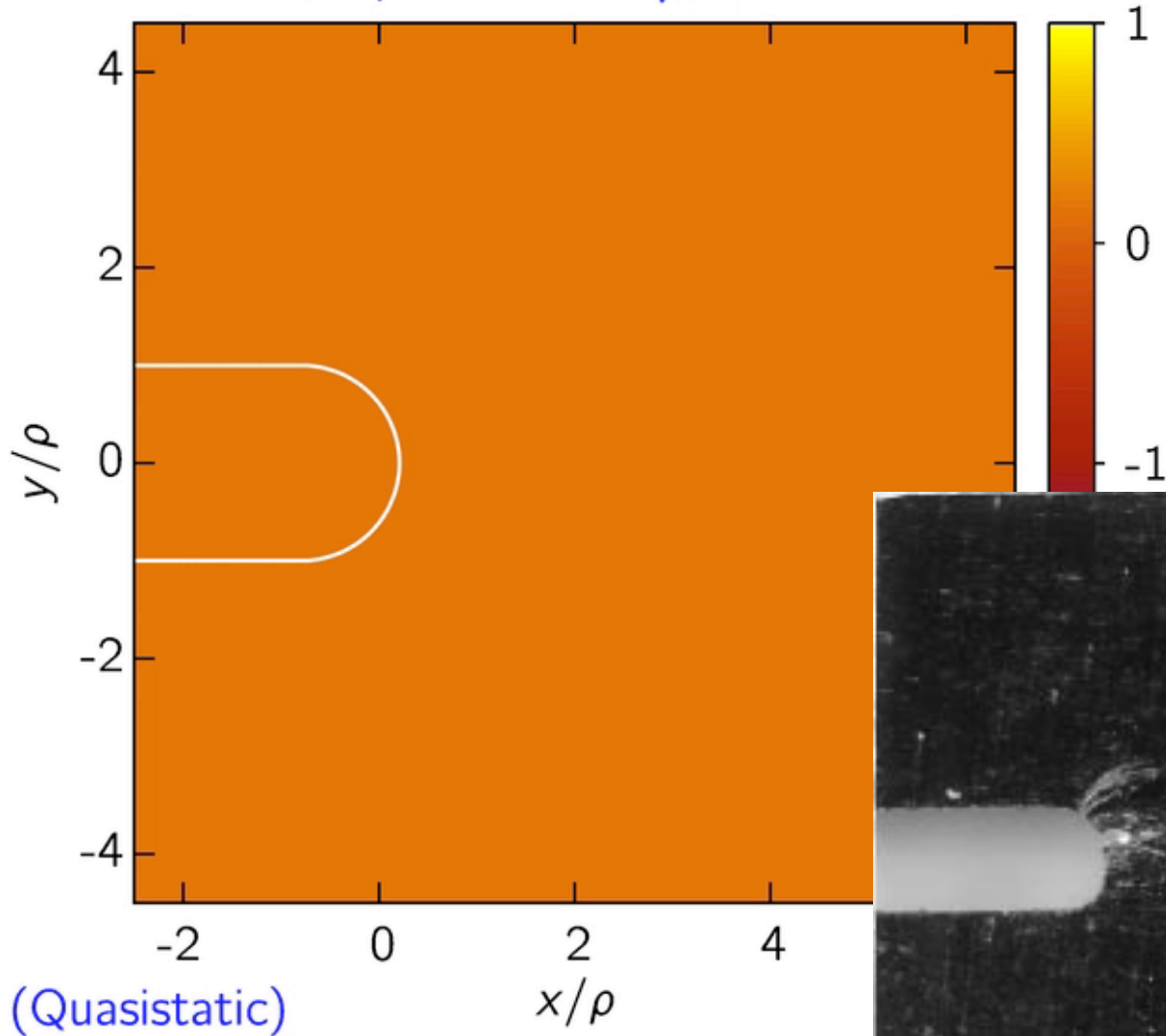
Chris Rycroft, UC Berkeley



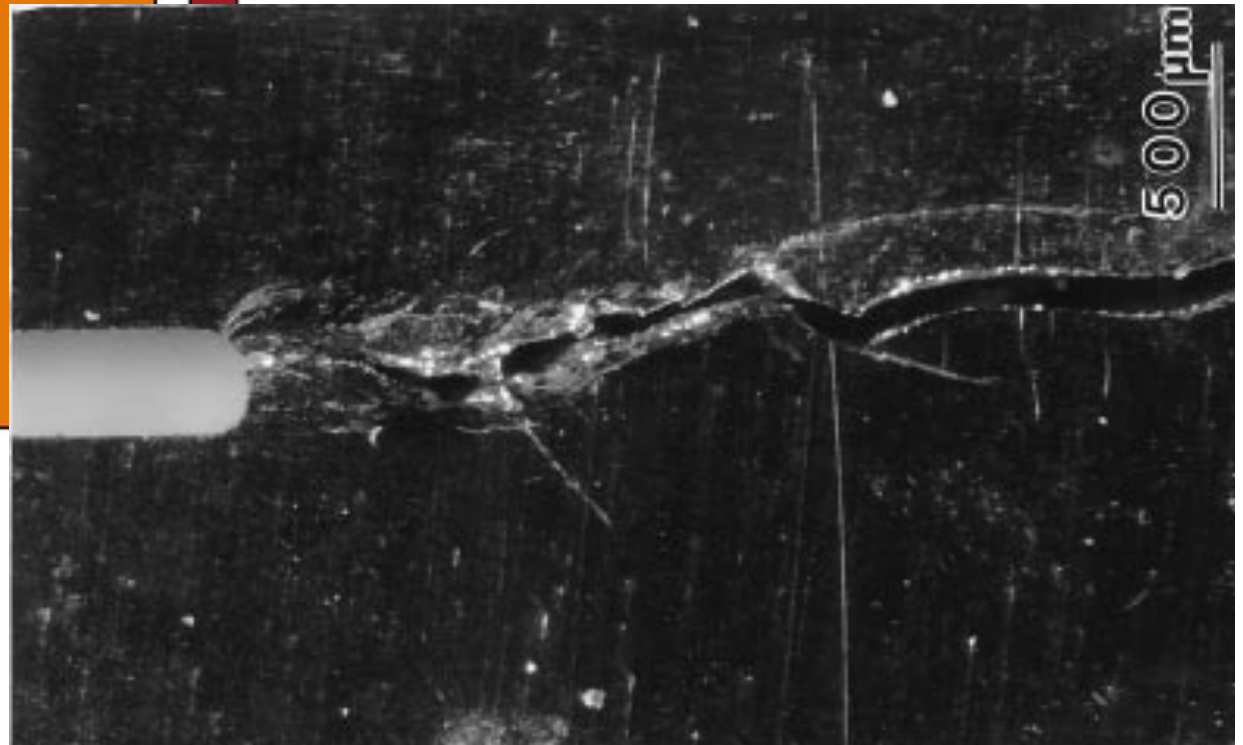
# Fracture with Cavitation

Rycroft, Bouchbinder PRL 109, 194301 (2012)

$t = 0 \text{ s}, K_I = 0 \text{ MPa } \sqrt{\text{m}}$

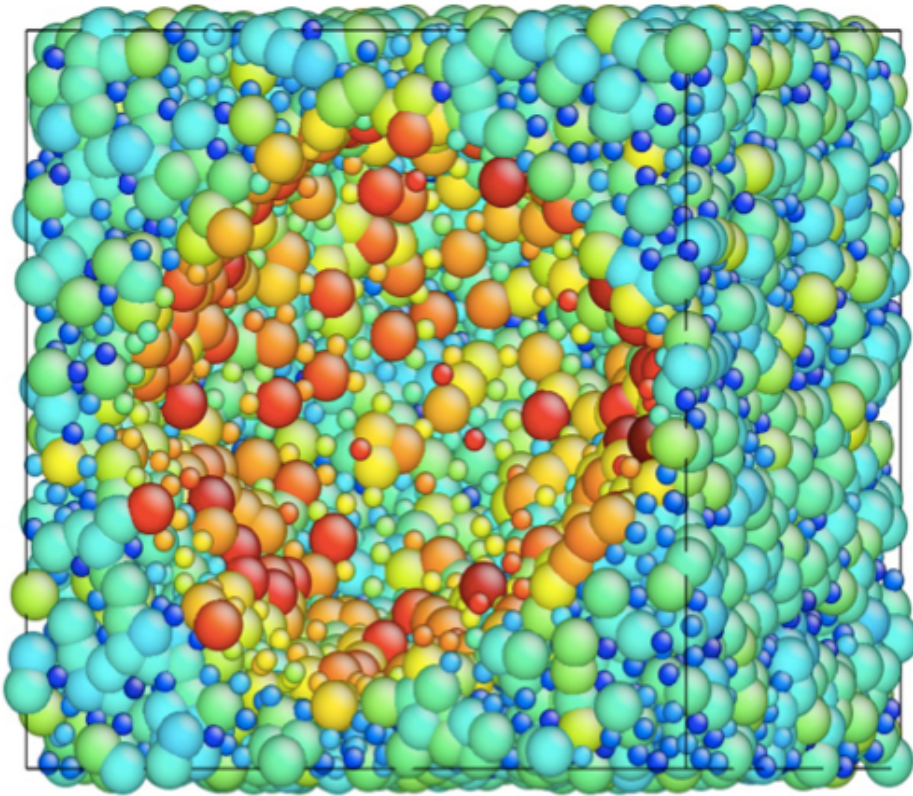


Lowhaphandu and Lewandowski  
Scripta Materialia 38, 1811 (1998)



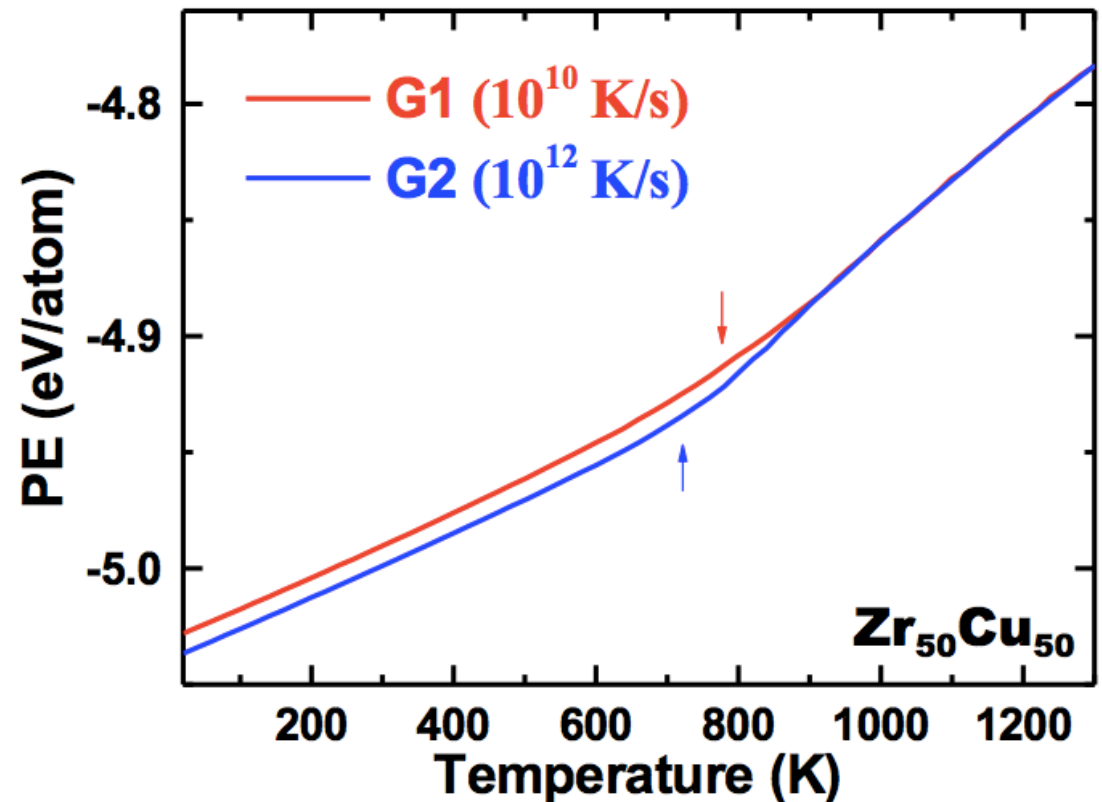


# Simulated System: Cu<sub>50</sub>Zr<sub>50</sub>

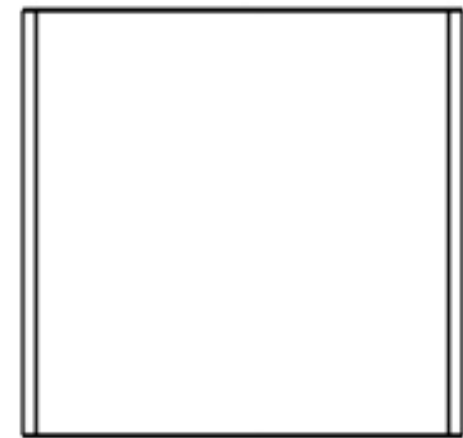
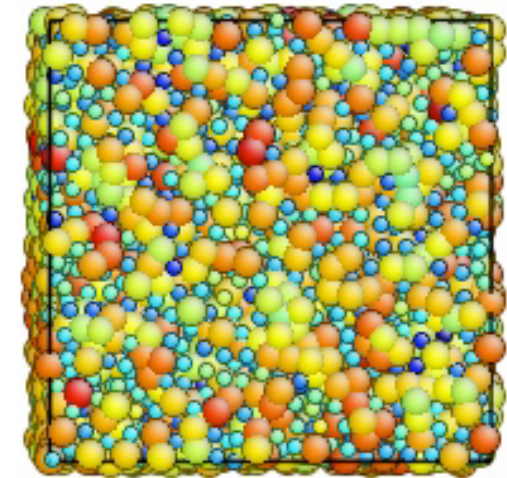
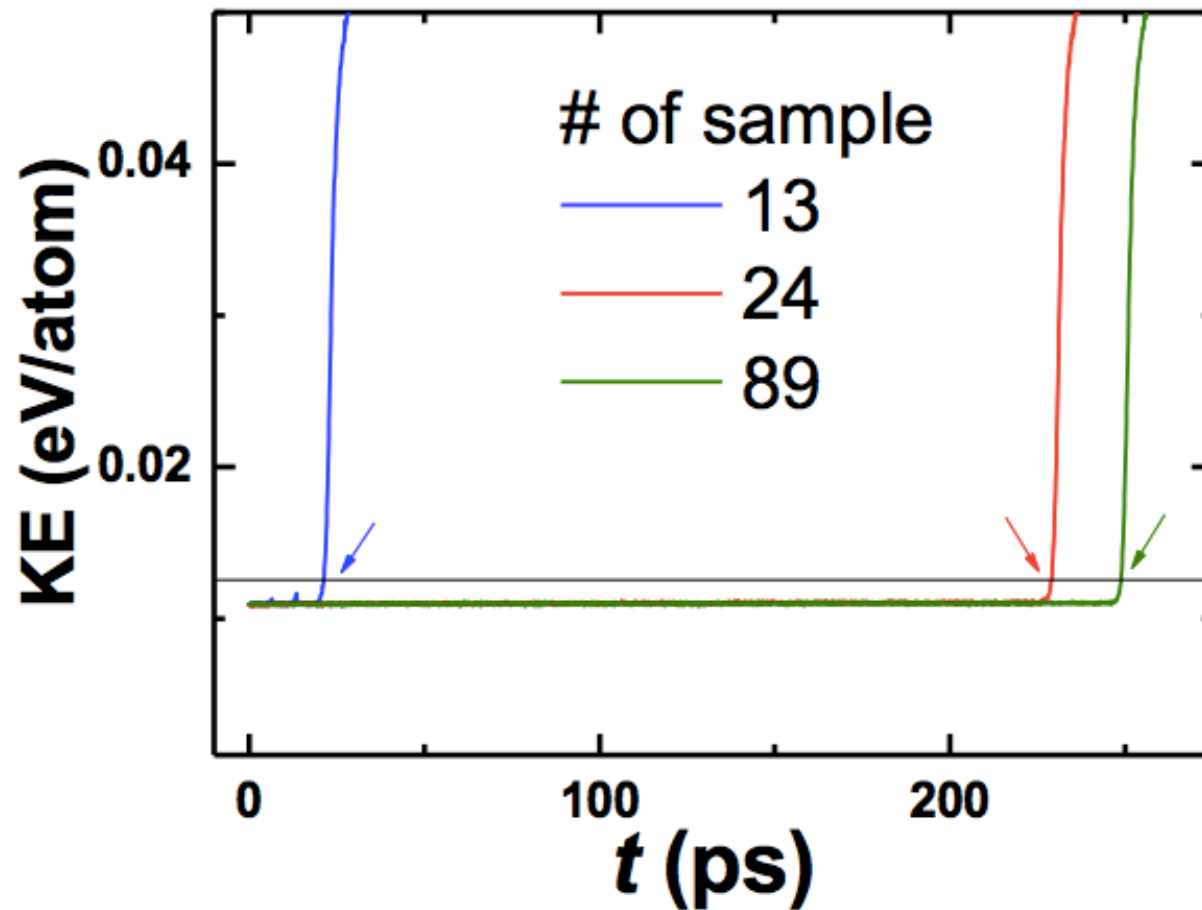


- ★ Unlike crystalline systems, it is not possible to skip simulating the processing step

- EAM Potential  
Y. Q. Cheng and E. Ma, Prog. Mater. Sci. 56, 379 (2011)
- 16,000 atoms
- Quench rates 1K/ps and 0.01K/ps



# Cavitation At Constant Strain





# Homogeneous or Heterogeneous?

- Macroscopically glasses appear isotropic and homogeneous.
  - Structurally glasses are in many ways structurally similar to a liquid.
  - However, cavitation in structurally identical samples with randomized initial velocities occurs at the same location implying heterogeneous nucleation.
- ➡ Use MD to test theories of nucleation.
- ➡ Measure the critical cavity size and determine how it compares to the CNT prediction.

# Classical Nucleation in Solids

- For a cavity nucleus to be unstable, the incremental elastic energy released must exceed the incremental surface energy and plastic work required to grow.

$$dF_e > dF_s + \delta W_p$$

$$4\pi K \epsilon_\infty r^2 dr > 8\pi \gamma r dr + \delta W_p(r, \dot{r})$$

- An estimate of the plastic work depends on assumptions regarding constitutive response, thermal softening and other difficult to ascertain details.

# Classical Nucleation in Solids

- We make 3 assumptions:
  - spherical cavity
  - rate independence of plastic work
  - plastic work scales with volume

$$4\pi K \epsilon_{\infty} r^2 dr > 8\pi \gamma r dr + \delta W_p(r, \dot{r})$$

- Under these conditions plastic work can be captured by a single parameter that determines the critical strain below which cavitation is impossible

$$4\pi K (\epsilon_{\infty} - \epsilon_c) r^2 dr > 8\pi \gamma r dr$$

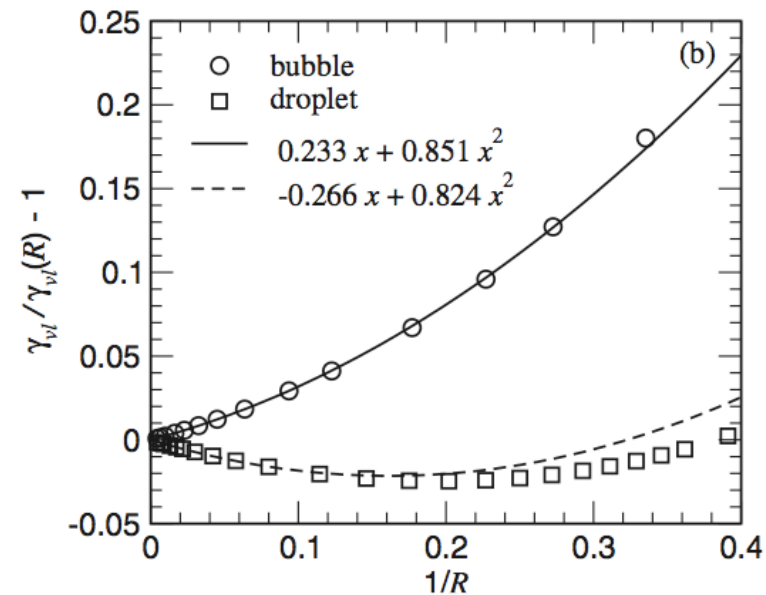
# Radius Dependence of Surface Energy

## "Tolman Length"

- It is known that for small nuclei (bubbles) the surface energy varies with bubble size

$$\gamma = \frac{\gamma_{\infty}}{1 + \frac{2\delta}{r}}$$

For work on Tolman length in liquid see  
An, Garrett, Samwer, Liu, Zybin, Luo, Johnson,  
Goddard, *Journal of Physical Chemistry*  
*Letters*, 2, 1320-1323 (2011)



Block, Das, Oettel, Viranu and Binder,  
*Curvature dependence of surface free energy of liquid drops*  
*and bubbles: A simulation study*, JCP, 133, 154702 (2010)



# Classical Nucleation in Solids

$$\gamma = \frac{\gamma_{\infty}}{1 + \frac{2\delta}{r}}$$

- We can infer a critical cavity size

$$r^* = \frac{2\gamma_{\infty}}{K(\epsilon - \epsilon_c)} \quad r_c = r^* \left( \frac{1}{2} + \frac{1}{2} \sqrt{1 + \frac{4\delta}{r^*}} - \frac{2\delta}{r^*} \right)$$

- And a critical free energy fluctuation for nucleation

$$F_c \equiv 4\pi\gamma r_c^2 - \frac{4}{3}\pi K(\epsilon_{\infty} - \epsilon_c)r_c^3$$

# Parameters in Theory

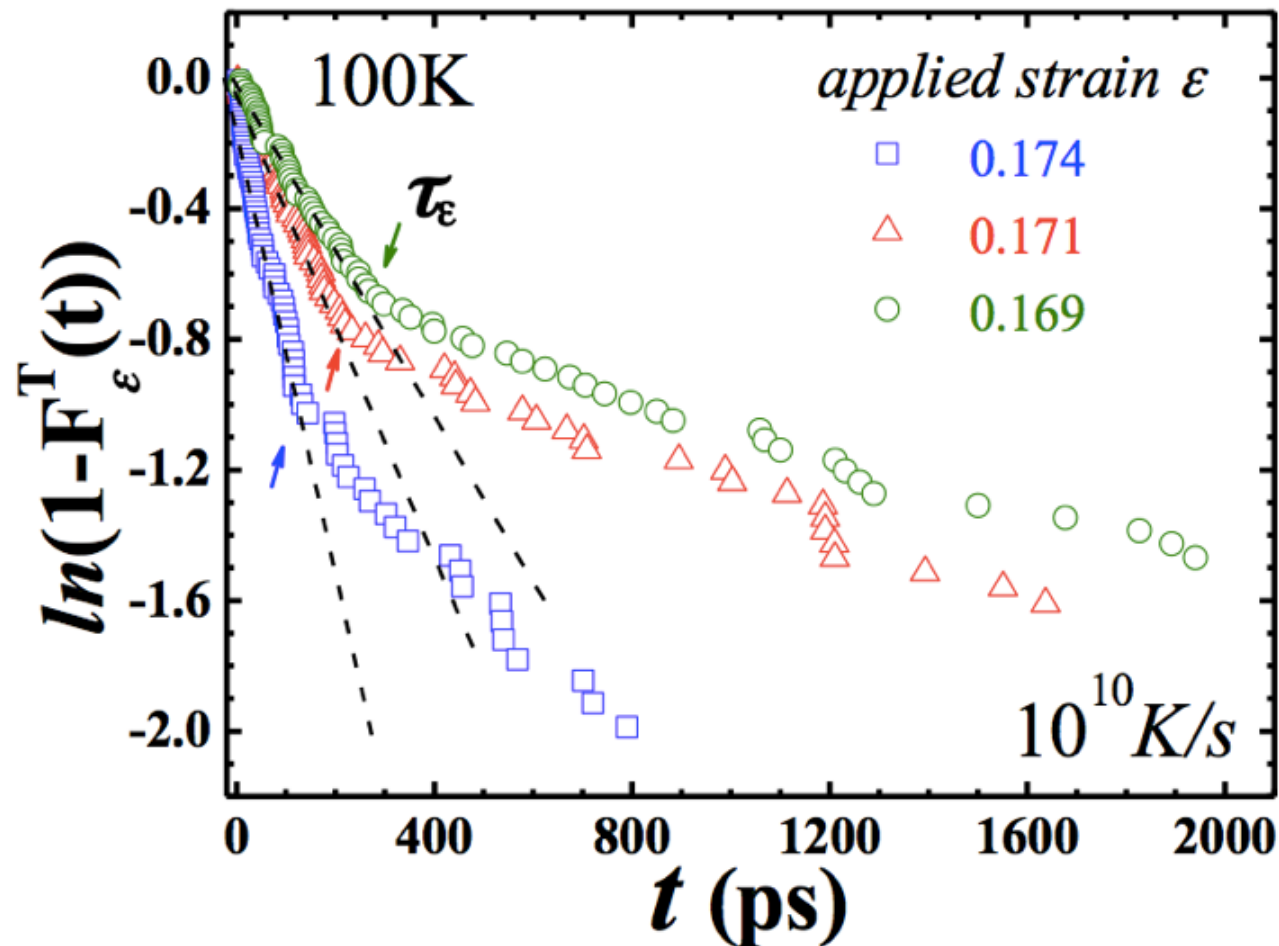
$$R = f \exp \left( -\frac{F_c}{k_B T} \right) \quad F_c \equiv 4\pi\gamma r_c^2 - \frac{4}{3}\pi K(\epsilon_\infty - \epsilon_c)r_c^3$$

$$r^* = \frac{2\gamma_\infty}{K(\epsilon - \epsilon_c)} \quad r_c = r^* \left( \frac{1}{2} + \frac{1}{2} \sqrt{1 + \frac{4\delta}{r^*}} - \frac{2\delta}{r^*} \right)$$

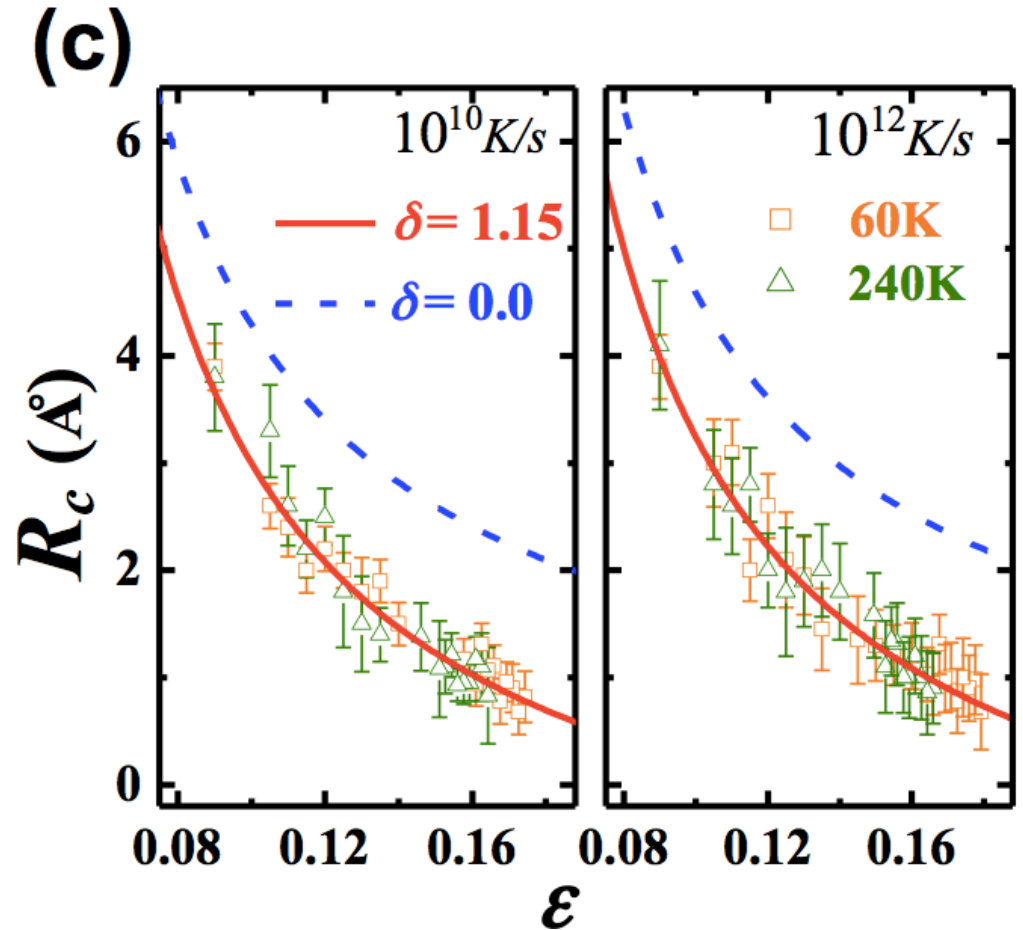
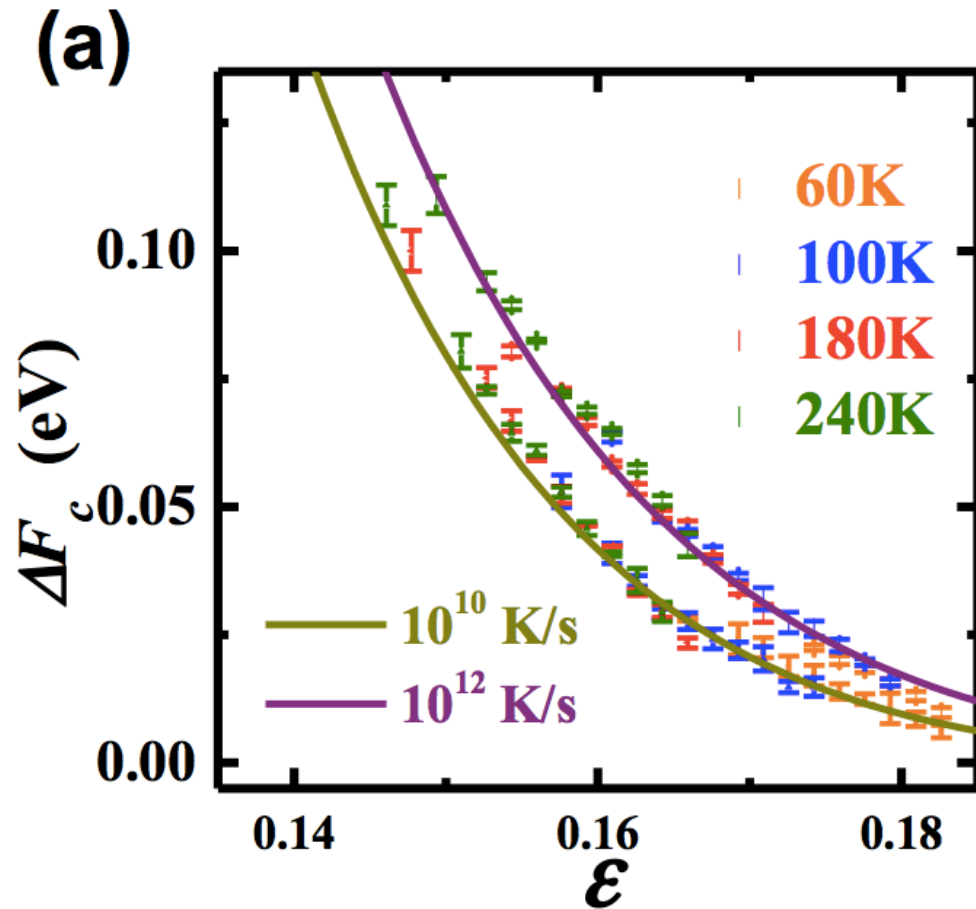
- Independent variables  $k_B T, \epsilon_\infty$
- Directly measurable parameters  $\gamma_\infty = 0.080 \text{ eV/\AA}^2$   
 $K = 106 \text{ GPa}$
- From T scaling  $f = 0.1 \text{ ps}^{-1}$
- From dKE/d(PV) during expansion  $\epsilon_c = 0.02$
- Tolman Length is fitted  $\delta = 1.15 \text{ \AA}$

# Cavitation: Strain and Hold

Log of fraction of samples not yet cavitared at a given time for a given applied remote strain.

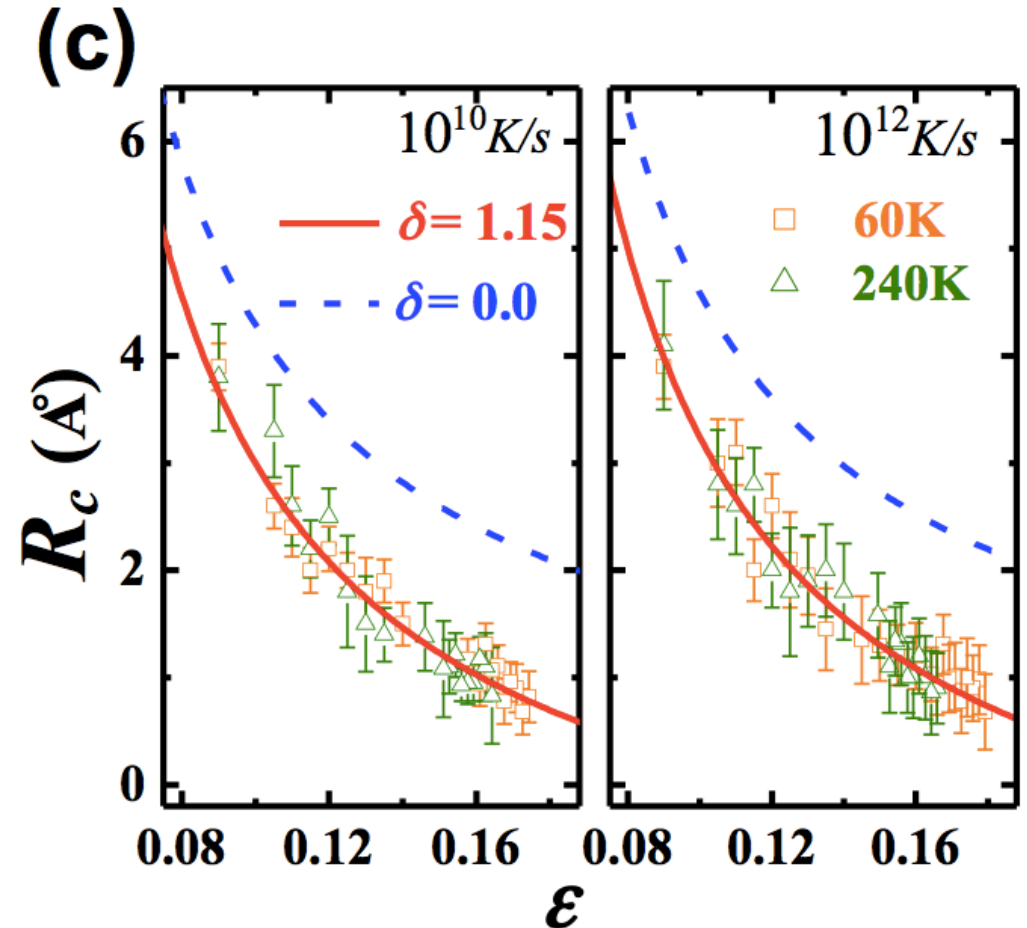
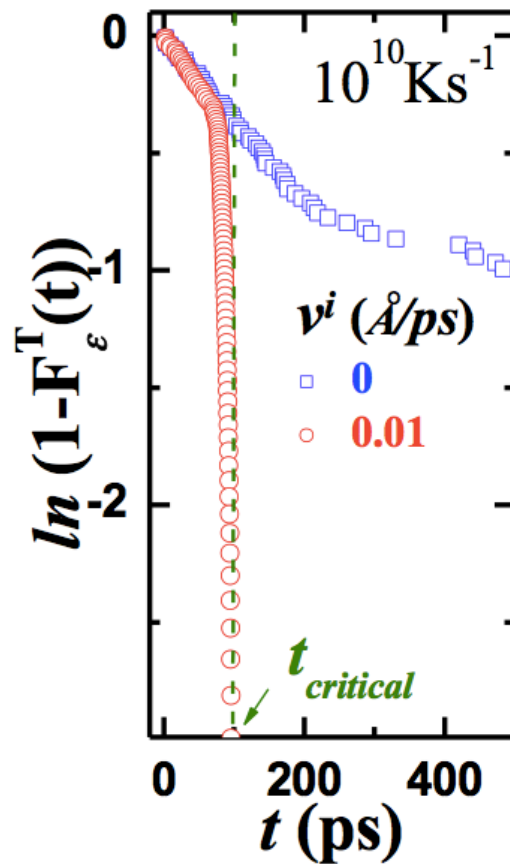
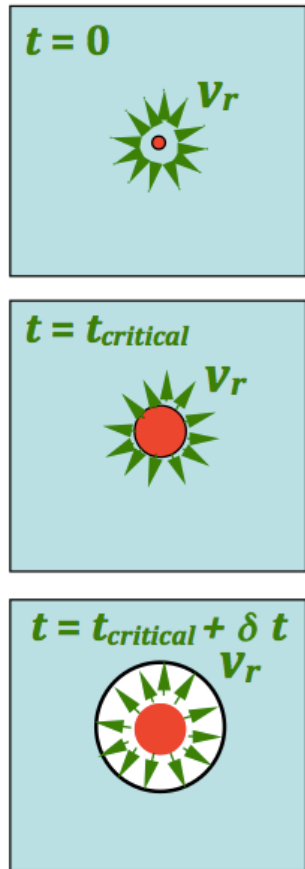


# Theory vs. Simulation: Prior to Aging

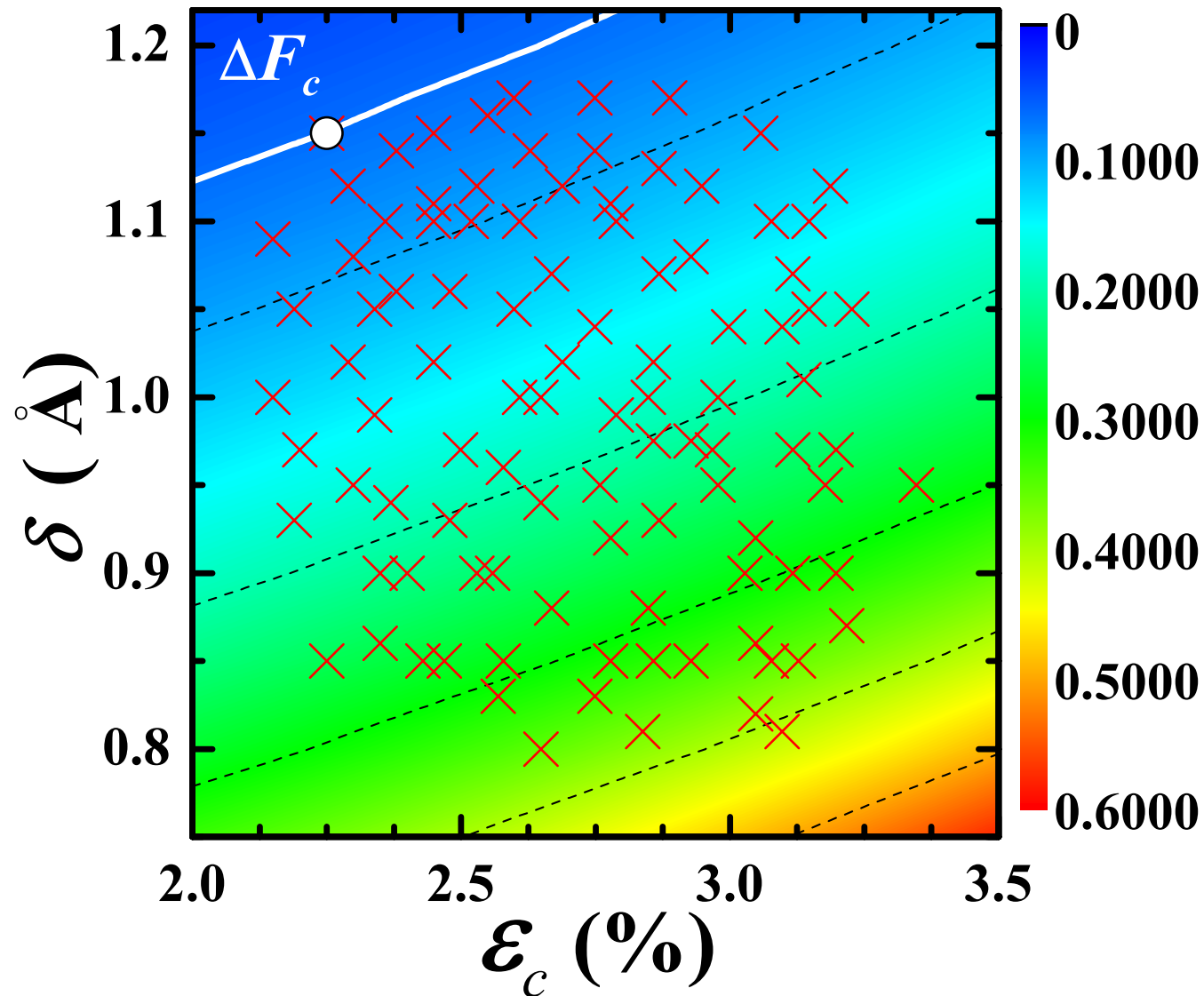




# Theory vs. Simulation: Prior to Aging

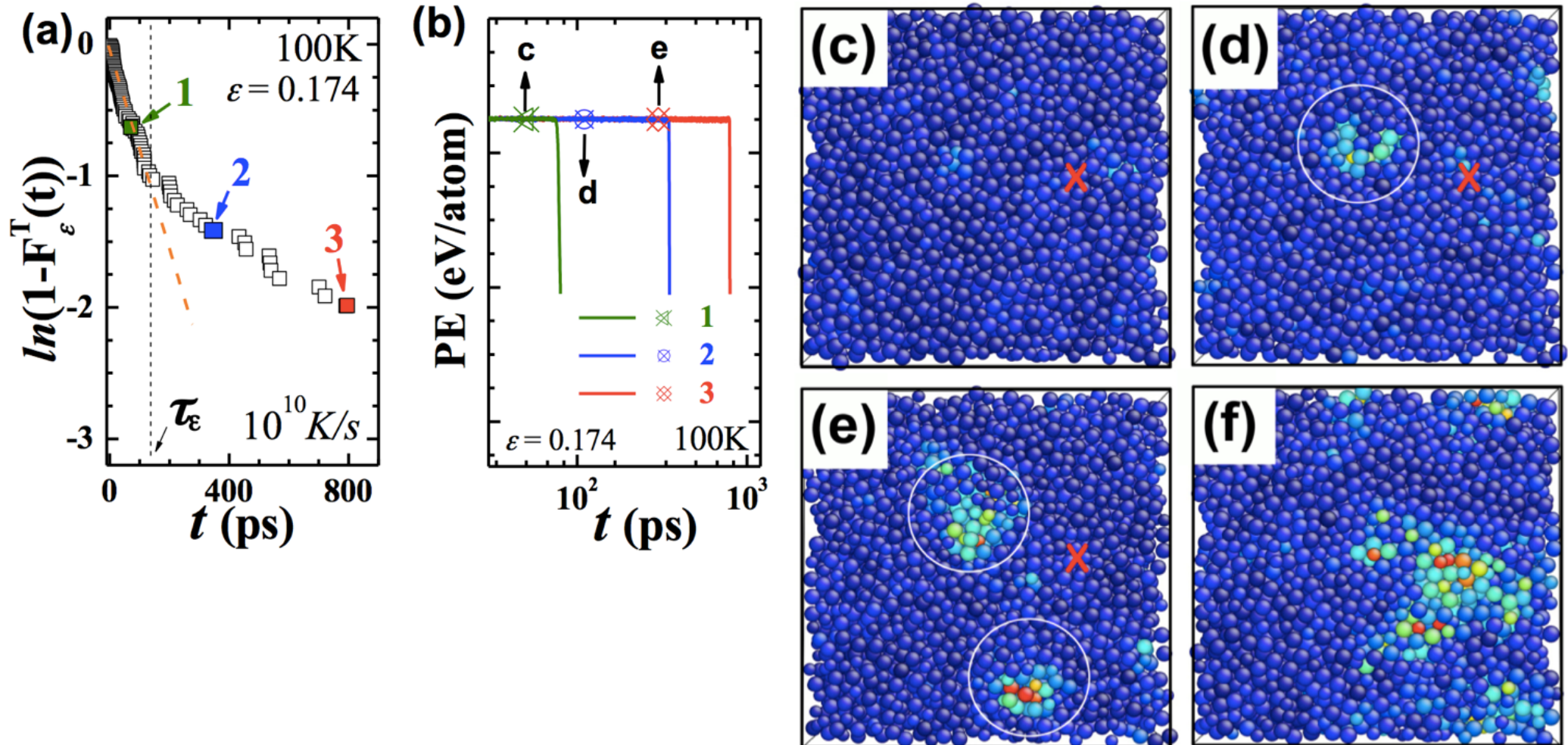


# Nature of Local Fluctuations

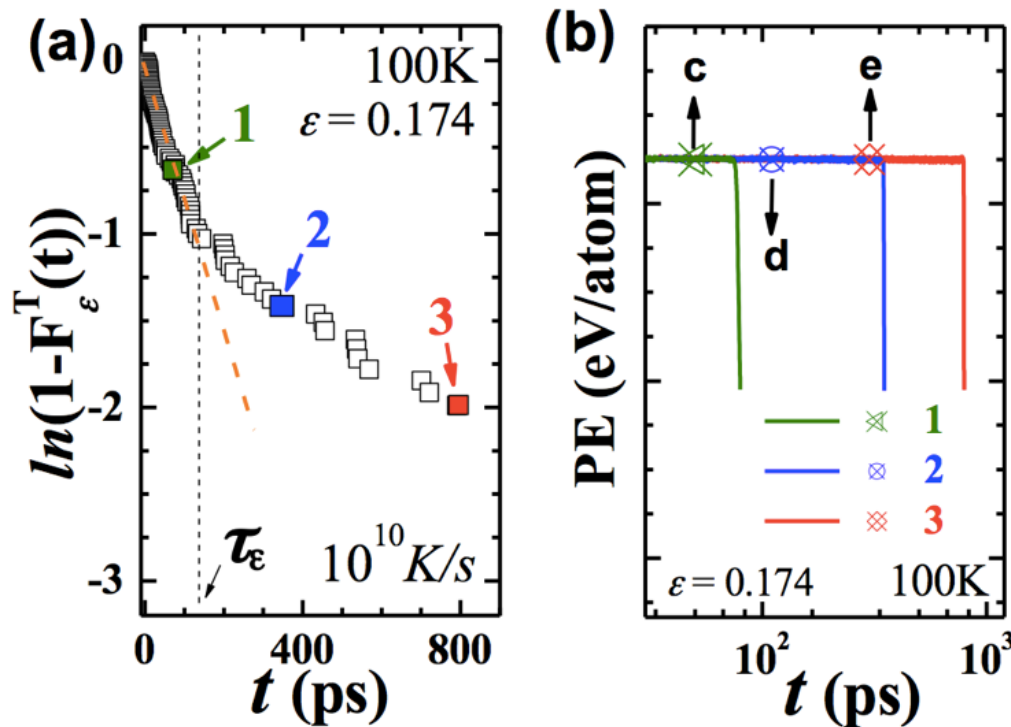


- Low correlation between Tolman length and critical strain ( $<0.13$ )
- 3.4 times more sensitive to Tolman length than critical strain.

# Onset of Aging: STZ activity

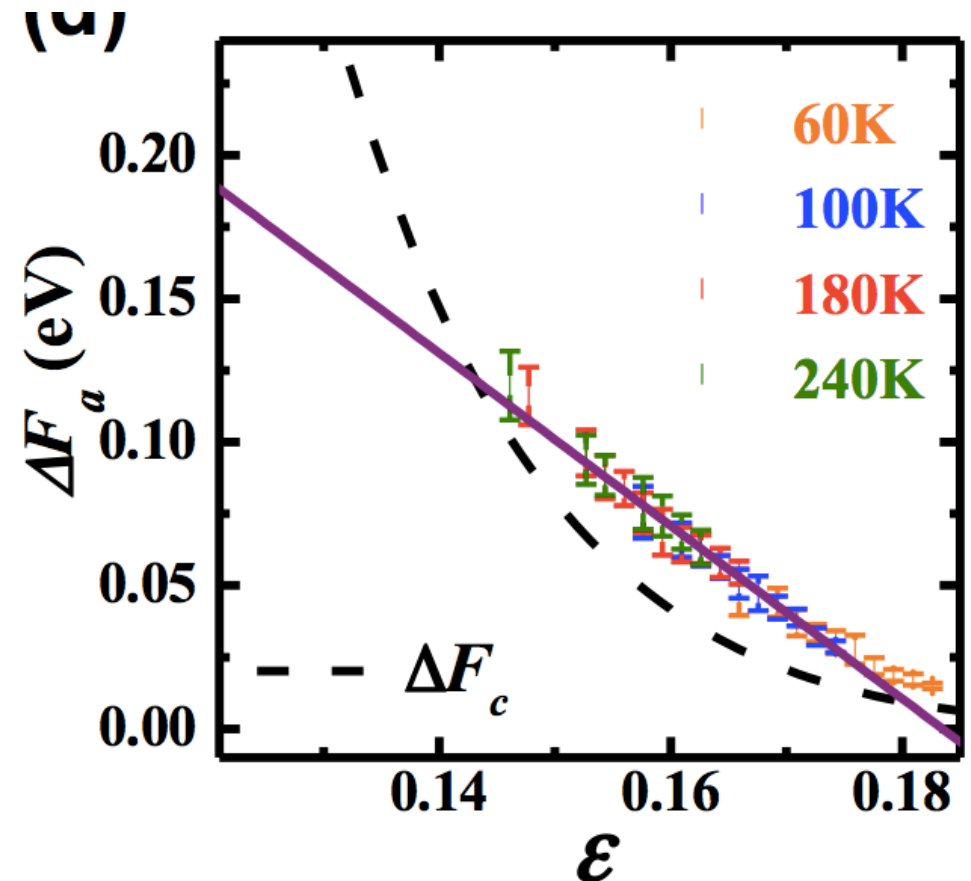


# Onset of Aging: STZ activity



- Measured activation volume:  $4.5\text{\AA}^3$
- Extrapolated zero strain activation energy: 0.55eV

$$R_a = f_a \exp \left( \frac{-\Delta F_a}{k_B T} \right)$$





# Conclusions

- Cavitation may play a crucial role in the fracture process zone.
- Plastic dissipation suppresses cavitation at low strains.
- Barrier to cavitation is reduced due to "Tolman Length" effects that decrease surface energy at small radii.
- At short time scales Poissonian rate is predicted by CNT.
- The critical radius under high tensile strain 14%-18% is very small (1Å-4Å) as confirmed by comparison to simulation.
- Spatial fluctuations in surface energy and, to a lesser degree, susceptibility to plasticity result in heterogeneous nucleation.
- Aging coincides with STZ activity for which we extract an activation barrier and activation volume. **postdoc opportunity!**
- STZ activation is favored over cavitation at low strains.
- Acknowledgement: NSF DMR 0808704 and 1107838